Simple Organometallic Chiral Derivatising Agents for the 31P N.M.R. Assay of the Enantiomeric Purity of Certain η^2 **-Donors**

David Parker* and Richard J. Taylor

Department of Chemistry, University of Durham, South Road, Durham DHI 3LE, U.K.

The chiral palladium and platinum ethene complexes **(1)** act as chiral derivatising agents for the **31P** n.m.r. assay of the enantiomeric purity of certain chiral alkenes and allenes.

With the heightened interest in asymmetric synthesis that pervades modern organic chemistry, there is a strong demand for further non-chiroptical methods for determining enantiomeric composition. Few such methods exist for assaying chiral alkenes, allenes, or alkynes, although advances have been reported in the application of chiral silver shift reagents, l and some work has been carried out with chiral stationary phases for gas chromatography.2 In seeking a simple n.m.r. method, we have taken advantage of the large chemical shift dispersion of 31P n.m.r.3 and the relative ease of displacement of ethene by other η^2 -donors in organometallic complexes.⁴ We report the use of the C_2 -symmetric ethene complexes (1) ,⁵ for the *in situ* 31P n.m.r. assay of the enantiomeric purity of certain chiral η^2 -donors.

The preparation and simple reactions of the desired (diop)- M^{0} -C₂H₄[†] complexes have been reported previously⁵ and the complexes may be stored at -20° C for months without decomposition. The ethene ligand **is** readily displaced by electron-poor alkenes. Reaction of **(la)** with a molar excess of the racemic acrylamide (2) in [²H₆]benzene led to four distinct AB quartets in the ³¹P n.m.r. spectrum (with associated platinum satellites) due to the four possible diastereoisomeric species. \ddagger These are formed by non-selective

 \sharp ³¹P N.m.r. data (C₆D₆, 298 K):

(i) and (ii) [and (iii) and (iv)] are unassigned constitutional isomers related by *Si-* or *Re-binding* of the alkene, while (i) and (iii) [and (ii) and (iv)] are diastereoisomers related by binding of the \hat{S} and \hat{R} enantiomers of **(2)** respectively by the same face.

 \dagger diop is 2,2-dimethyl-4,5-bis(diphenylphosphinomethyl)-1,3-dioxolane.

binding of the *Si* and *Re* face of both enantiomers. Using the enantiomerically pure *R* and *S* acrylamides, mixtures of known enantiomeric composition were made up and assayed. **A** linear plot of enantiomeric composition *vs.* **31P** n.m.r. measured values was obtained, consistent with no enantioselectivity in binding. Such behaviour may be expected when the chiral centre under examination is remote from the metal centre. Similar behaviour was observed with the diastereoisomeric complexes formed by reaction of (1a) and **(1b)** with (R) -or (S) -5-isopropenyl-2-methylcyclohex-2-enone (carvone), **(3).** In this case, however, the metal is bound with complete stereoselectivity to the less hindered *Si-Si* face of the endocyclic, electron-poor double bond. Typical spectra for the diastereoisomeric platinum complexes are shown in Figure 1, and there was no evidence for selective complexation of one enantiomer. Moreover, the **31P** n.m.r. method permits a precise determination of enantiomeric purity. **A** commercial sample of $(-)$ -carvone (Fluka 22060) was 96% (\pm 1%) *R*, while the enantiomer $(+)$ -carvone was determined to be $>99\%$ *S* (\pm 0.5%). A sample of (+)-carvone (Koch-Light) of indeterminate age was 86% **S.**

The co-ordinated ethene ligand in **(1)** is also rapidly displaced by addition of allenes and strained alkenes. Excess norbornene reacts with **(la)** or **(lb)** to give one diastereoisomer $[(b) \delta_P(C_6D_6) + 6.2 p.p.m., (a) \delta_P 15.8, J_{PtP} 3414 Hz]$, consistent with selective complexation of the *exo* face of the double bond. Reaction of **(la)** with racemic dimethyl *trans***norbornene-2,3-dicarboxylate (4)** gave two diastereoisomeric species, precisely in the ratio *50:50.* Reaction of **(la)** with 1,1-dimethylallene gave a single complex (δ_P 18.5, 6.3 p.p.m.; $J_{\text{PtP(1)}}$ 3473, $J_{\text{PtP(2)}}$ 2842, $J_{\text{P(1)}(2)}$ 55 Hz) with preferential binding of the more substituted double bond,⁵ while racemic cyclonona-1 ,2-diene *(5)* was selectively bound by the lesshindered face to give two diastereoisomers in equal ratio [(i) $\delta_{P(1)}$, 17.6, $\delta_{P(2)}$ 11.0 p.p.m., $J_{PtP(1)}$ 3246, $J_{PtP(2)}$ 3057, $J_{P(1)P(2)}$ 71 Hz, (ii) $\delta_{P(1)}$ 17.3, $\delta_{P(2)}$ 10.3 p.p.m., $J_{PtP(1)}$ 3250, $J_{PtP(2)}$

Figure 1. 31P N.m.r. spectra $(C_6D_6, 298$ K, 101.3 MHz) of the diastereoisomeric complexes formed by reaction of **(la)** with different samples of carvone of varying enantiomeric composition. Data $(C_6D_6,$ 298 K): $[R-(1a)]-[S-(3)]$ $\delta_{P(1)}$ 12.5, $\delta_{P(2)}$ 10.8 p.p.m., $J_{PtP(1)}$ 3409, $J_{\text{PrP}(2)}$ 3881, $J_{\text{P}(1)\text{P}(2)}$ 65 Hz; $[R-(1a)]-[R-(3)]$ $\delta_{\text{P}(1)}$ 13.8, $\delta_{\text{P}(2)}$ 9.9 p.p.m.9 **JPtP(1)** 3537, **JPtP(Z)** 39387 **JP(,)P(2)** *65* €32.

3060, $J_{P(1)P(2)}$ 71 Hz). Using a literature method for the synthesis of enantiomerically enriched (5), a sample was assayed to be *5%* S.6

1

There are some obvious limitations to this **31P** n.m.r. technique. The co-ordinated ethene ligand in **(1)** is not displaced, at room temperature, by excess cyclohexene or cyclopentene or by simple alkyl-substituted ethenes. Furthermore, if the centre of chirality is very close to the metal centre or if a substituent may additionally bind to the metal, then preferred binding of one enantiomer negates the use of this method. For example, the chiral alkyne **(6)** reacted with **(la)** to give one diastereoisomer only $(\delta_{P(1)} 10.8, \delta_{P(2)} 4.6 \text{ p.p.m.})$ $J_{P(1)P(2)}$ 33, $J_{P(P(1))}$ 3769, $J_{P(P(2))}$ 3247 \overrightarrow{Hz}). Such behaviour is, of course, well known in studies relating to the mechanism of asymmetric catalysis, involving chiral organopalladium7 or organorhodium complexes *.8*

We thank the S.E.R.C. and Glaxo Group Research **(Greenford) for a CASE studentship and Dr. Philip J. Sidebottom for some helpful comments.**

5 Received, 6th July 1987; Corn. 953

References 7

- **¹**A. Mannschreck, W. Munninger, T. Burgemeister, J. Gore, and **B.** Cazes, *Tetrahedron,* **1986, 42, 399;** W. Offermann and **A.** Mannschreck, *Tetrahedron Lett.,* **1981, 22, 3227.**
- **2 V.** Schurig and E. Gi1.-Av., *Isr.* J. *Chem.,* **1977, 15, 96.**
- E. F. Mooney and G. **A.** Webb, *Annv. Rep. NMR Spectrosc.,* **1972, 58;** P. M. Cullis and G. Lowe, J. *Chem. SOC., Chem. Commun.,* **1981, 2317; B.** Feringa, **J.** *Chem. SOC., Chem. Commun.,* **1987, 695.**
- **4** *E.g.* D. Parker, J. *Orgqnomet. Chem.,* **1982, 240, 83.**
- **J.** M. Brown, **S.** J. Cook, and **S.** J. Kimber, J. *Organomet. Chem.,* **1984, 269, C58;** M. Hodgson and D. Parker, J. *Organomet. Chem.,* **1987,325, C27;** M. Hodgson, D. Parker, and R. J. Taylor, *I. Chem.* **SOC.,** *Chem. Commun.,* **1987, 1309.**
- **6** L. Skattelbol and **S.** Solomon. *Org. Synth.,* **1969, 49, 35;** H. Nozaki, **A.** Aratani, T. Toraya, and R. Noyori, *Tetrahedron,* **1971, 27, 905.**
- **P.** B. Mackenzie, J. Whelan, and B. Bosnich, *J. Am. Chem. SOC.,* **1985, 107, 2046.**
- **8 J.** M. Brown and D. Parker, J. *Org. Chem.,* **1982,47,2722; J.** M. Brown and **P.** A. Chaloner, J. *Chem. SOC., Perkin Trans.* 2, **1982, 711.**